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MEMORANDUM FOR PRS (Contractor/In-House Publication)

FROM: PROI (TI) (STINFO)

24 Oct 2000

SUBJECT: Authorization for Release of Technical Information, Control Number: **AFRL-PR-ED-TP-2000-205**
Suri, Suresh; Tinnierllo, M. & Marcischak, J. (ERC), "Synthesis and Screening of Advanced Hydrocarbon Fuels"

2000 USAF High Energy Density Matter (HEDM) Contractors Conference (Statement A)
(Park City, UT, 24-26 Oct 2000) (Deadline: PAST)

1. This request has been reviewed by the Foreign Disclosure Office for: a.) appropriateness of distribution statement, b.) military/national critical technology, c.) export controls or distribution restrictions, d.) appropriateness for release to a foreign nation, and e.) technical sensitivity and/or economic sensitivity.

Comments: _____

Signature _____ Date _____

2. This request has been reviewed by the Public Affairs Office for: a.) appropriateness for public release and/or b) possible higher headquarters review

Comments: _____

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Comments: _____

APPROVED/APPROVED AS AMENDED/DISAPPROVED

PHILIP A. KESSEL
Technical Advisor

Date

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Synthesis & Screening of Advanced Hydrocarbon Fuels

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Presentation Outline

- Goal
 - HEDM program
 - NASA program
 - IHPRPT program (propellant perspective)
- Criteria for fuel selection
- Approach
- Results
- Accomplishments (FY-2000)
- Planned Efforts (FY-2001)



HEDM Goal



- To Develop fuels with increased Isp over LOX/RP-1
 - LOX/RP-1 (Calculated Isp) = 300 sec
 - LOX/RP-1 (Delivered Isp) = 263 sec

Determined at sea level and 1000 psi chamber pressure



IHPRPT GOAL (Propellant Contribution)

To Meet IHPRPT Phase II and Phase III Objective

Phase	Time	Improvement Over SOTA* Isp (del)
II	2005	+ 5 Sec
III	2010	+ 11 Sec

*SOTA: LOX/RP-1 Propellant Isp(del) = 263 Sec.
Isp (calc) = 300 Sec.

NASA Goal



- **FY-1999**
 - Deliver three advanced hydrocarbon fuel in 8-10 lb quantity.
 - Quadricyclane
 - 1,7-Octadiyne
 - Bicyclopolydene
- **FY-2000**
 - Screen four hydrocarbons for their physical and hazardous properties.

Criteria for Fuel Selection



- Predicts better performance (Isp) over LOX/RP-1 system
- Most desirable physical properties
 - Lower vapor pressure compared to RP-1
 - Higher density (\geq RP-1 = 0.801 g/ml)
 - Freezing point (\leq -10 °C; RP-1 = -41.4 °C)
 - Boiling point \geq B. P. Of RP-1
- Storable
- Compatible with the current system

Approach



- Structural requirements
- Survey of energetic hydrocarbons
- Selection of hydrocarbons based on improved theoretical performance
- Synthesis of target hydrocarbons at bench scale.
 - *Easy preparation, cost effective and safe*
- Translate bench-scale synthesis to pilot scale.

Heat of Formation of Saturated Hydrocarbons



Compound	Structure	ΔH_f (Obs)
Ethane	CH_3CH_3	-20.04
Propane	$\text{CH}_3\text{CH}_2\text{CH}_3$	-25.02
Butane	$\text{CH}_3(\text{CH}_2)_2\text{CH}_3$	-30.03
Pentane	$\text{CH}_3(\text{CH}_2)_3\text{CH}_3$	-35.08
	$\Delta H_f/\text{added } \text{CH}_2 = \sim -5 \text{ Kcal/mole}$	

Heat of Formation of Unsaturated Hydrocarbons



Compound	Structure	$\Delta H_f(\text{Obs})$
• Ethylene	$\text{CH}_2=\text{CH}_2$	+12.5
• 1,3-Butadiene	$\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2$	+26.11
•	$\Delta H_f/\text{C} = \sim +6.25 \text{ Kcal/mole}$	
• Acetylene	$\text{HC}\equiv\text{CH}$	+54.36
•	$\Delta H_f/\text{C} = \sim +27.1 \text{ Kcal/mole}$	



Structural Requirement for High Energy Contents (Cont..)

- The energy content is also increased by incorporating strain in the molecule
 - Ring compound ΔH_f + 12.73 kcal/mole
 - Cyclopropane + 6.78 kcal/mole
 - Cyclobutane - 18.44 kcal/mole
 - Cyclopentane

Structural Requirements For High Energy Contents (Summary)

Incorporation of small ring (strain) and unsaturation in a molecule increases its energy contents



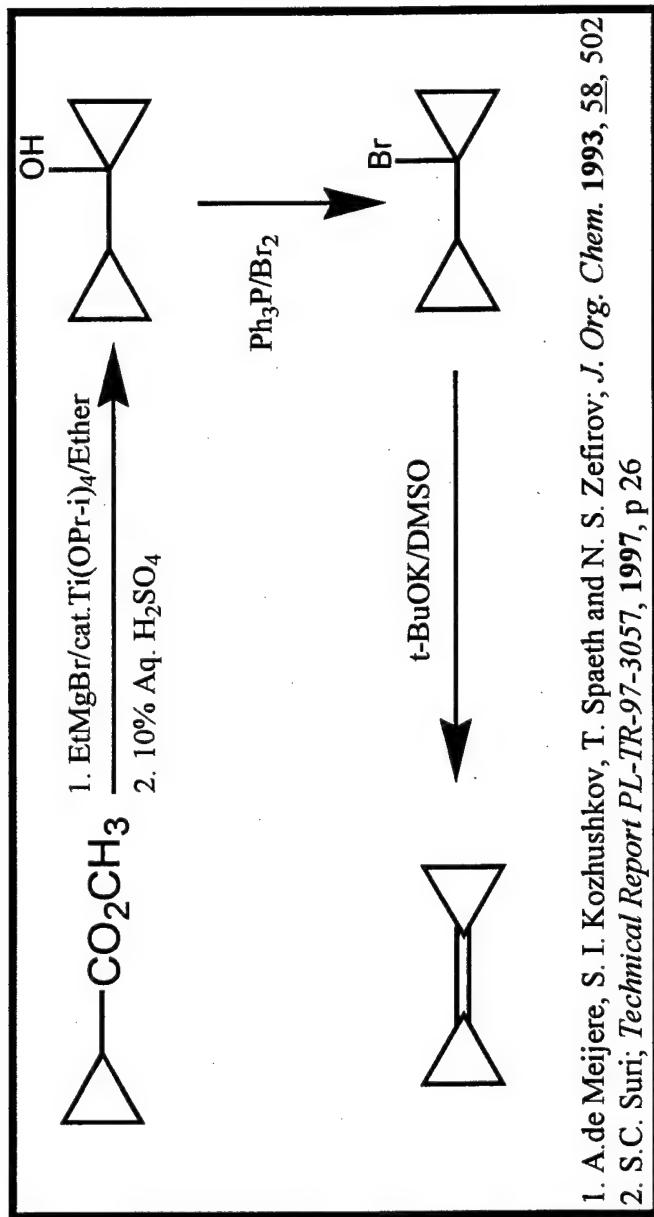
Performance Comparison of Energetic Hydrocarbons (Theoretical)



Hydrocarbons	H/C ratio	Density (g/ml)	Calc. ΔHf (Kcal/mole)	Calc. Isp (sec)
RP-1	1.9	0.80	-5.76	300.0
Quad	1.14	0.98	72.2	307.0
BCP	1.33	0.85	76.1	312.5
AFRL-1	1.2	0.77	64.0	311.3
AFRL-2	1.25	0.87	73.4	307.2
AFRL-3	1.0	0.93	123.6	307.2
AFRL-4	1.0	-	129.6	321.4
AFRL-5	1.33	0.80	56.3	308.7

Results

Synthetic Sequence of BCP



1. A. de Meijere, S. I. Kozhushkov, T. Spaeth and N. S. Zefirov; *J. Org. Chem.* 1993, **58**, 502
2. S.C. Suri; *Technical Report PL-TR-97-3057*, 1997, p 26





Characterization of BCP

Physical properties

B.P. = 101 °C

M.P. = -12 °C

F.P. = -6.4 °C

Density = 0.8454 g/ml

ΔH_f (exp.) = 67.4 kcal/mole

ΔH_f (calc.) = 76.1 kcal/mole

Hazardous properties

Zero card gap (negative)

Drop test > 200 kg/cm

Friction test 133 newton

Toxicity

(Inhalation LC50)

1.95 mg/L

Adiabatic Compression(ksi)

3000	Neg.
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Is BCP Hypergolic?



- Qualitative Test

- BCP is found to be hypergolic using nitrogen tetroxide(NTO). Spontaneous reaction with visible flame.
- Hypergolic with inhibited red fuming nitric acid (IRFNA) as oxidizers. (Darren M. Thompson, U.S. Army missile command).
- Ignition Delay
 - The work is in progress under SBIR phase-1 with TDA Research, Inc.

Synthesis of AFRL-1

- Two steps synthesis
- Involves readily available materials
- Yield in both steps is $> 90\%$





Characterization of AFRL-1

Physical Properties

B.P. = 52- 55 °C

Density = 0.77 g/ml

ΔH_f (Exp.) = 67.4 Kcal/mole

ΔH_f (Calc.) = 64.0 Kcal/mole

Hazardous Properties

“0” card gap (Negative)

Liq. Impact test > 200 Kg-cm

Friction Test 78 Newtons

Adiabatic Compression (psi)

3000	Neg.
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Synthesis of AFRL-3

- One step synthesis from AFRL-1.
- Requires oxidative coupling of AFRL-1.
- Yield is 92 %.





Characterization of AFRL-3

Physical Properties

B.P. = 102 °C

M.P. = -13 °C

Density = 0.93 g/ml

ΔH_f (Calc.) = 123.6 kcal/mole

ΔH_f (Exp.) = 117.0 kcal/mole

Hazardous Properties

- “0” card gap (negative)
- Liq Impact test <20 kg-cm
- Friction Test = 64.8 Newton

Adiabatic Compression(ksi)

500	Neg.
2000	Neg.
3000	Neg.

Synthesis of AFRL-5



- Higher homologue of AFRL-1
- Two step synthesis
- Yield in both steps is greater than 90 %



Characterization of AFRL-5

Physical properties

B.P. = 78 °C
M.P. = -92.8 °C
Density = 0.7957 g/ml
 ΔH_f (Exp.) = 50.39
kcal/mole
 ΔH_f (Calc.) = 56.3 kcal/mole

Hazardous properties

“0” card gap (TBD)
Liq. Impact Test > 200 kg-cm
Friction Test = 43.12 newton
Adiabatic Compression (psi)

3000	Neg.
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Accomplishments (FY 00)



- Delivered four hydrocarbons to NASA/Marshall.
 - Cyclopropyl acetylene (AFRL-1).
 - Bicyclopropylidene
 - Quadricyclane
 - 1,7-Octadiyne
- Synthesized two advanced hydrocarbons (AFRL-1 & AFRL-3) at bench-scale level.
- 200 gm of AFRL-3 was synthesized in the laboratory.

Planned Efforts of Fiscal Year 2001 (Technical)



- To continue exploring bench scale synthesis of advanced hydrocarbon (AFRL-4).
- Evaluate physical & hazardous properties of AFRL-4 & AFRL-2.



Alliances

- **Industry**
 - Boeing
 - TRW
 - Kistler
 - Aerojet
- **NASA**
 - Marshall
 - Glenn
- **DOD**
 - Navy- China Lake
 - Army- Huntsville

Team Efforts



Research

- Suresh C. Suri
Michael Tinnirello
Jacob Marcischak

Theoretical Efforts

- Jeffrey Mills

Physical Properties

- Paul Jones, JoAnne Larue,
Jeff Yinn

Hazardous Properties

- Tommy W. Hawkins,
Adam Brand, Milton
Mckay, Ismail Ismail

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